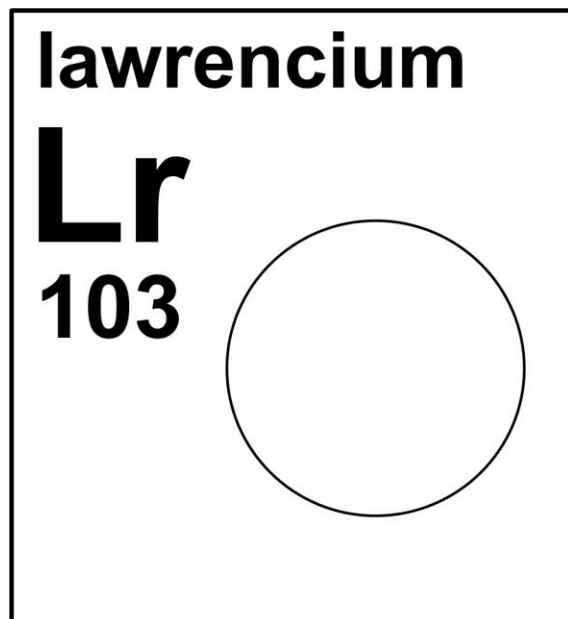





lawrencium



Stable isotope	Atomic mass	Mole fraction
(none)		

Half-life of radioactive isotope

Less than 1 second 
 Between 1 second and 1 hour 
 Greater than 1 hour 

$^{251}_{\text{Lr}}$	$^{252}_{\text{Lr}}$	$^{253}_{\text{Lr}}$	$^{254}_{\text{Lr}}$	$^{255}_{\text{Lr}}$	$^{256}_{\text{Lr}}$	$^{257}_{\text{Lr}}$	$^{258}_{\text{Lr}}$	$^{259}_{\text{Lr}}$	$^{260}_{\text{Lr}}$
$^{261}_{\text{Lr}}$	$^{262}_{\text{Lr}}$	$^{263}_{\text{Lr}}$	$^{264}_{\text{Lr}}$	$^{265}_{\text{Lr}}$	$^{266}_{\text{Lr}}$				

Important applications of stable and/or radioactive isotopes

Applications: Lawrencium has no known isotopic applications aside from scientific research.